

Faster Decomposition Under Increased Atmospheric CO₂ Limits Soil Carbon Storage

Kees Jan van Groenigen,^{1,2*} Xuan Qi,³ Craig W. Osenberg,⁴ Yiqi Luo,^{3,5} Bruce A. Hungate^{1,6}

Soils contain the largest pool of terrestrial organic carbon (C) and are a major source of atmospheric carbon dioxide (CO₂). Thus, they may play a key role in modulating climate change. Rising atmospheric CO₂ is expected to stimulate plant growth and soil C input but may also alter microbial decomposition. The combined effect of these responses on long-term C storage is unclear. Combining meta-analysis with data assimilation, we show that atmospheric CO₂ enrichment stimulates both the input (+19.8%) and the turnover of C in soil (+16.5%). The increase in soil C turnover with rising CO₂ leads to lower equilibrium soil C stocks than expected from the rise in soil C input alone, indicating that it is a general mechanism limiting C accumulation in soil.

Carbon exchange between land and the atmosphere is a major focus of Earth system models, because the C cycle is sensitive to environmental change and because alterations of the C cycle affect our climate. Earth system models project that rising atmospheric CO₂ will promote C uptake by the terrestrial biosphere (1). The resulting increase in C stocks in plant biomass and soil organic matter (2) would slow the rise in atmospheric CO₂ concentrations and may help to slow climate change. Data support part of these model predictions: Experimentally increased CO₂ concentrations usually stimulate photosynthesis and plant growth (3). However, the response of soil C stocks is less well understood, because changes in soil C content are difficult to detect (4, 5). Thus, global

models projecting future C dynamics of the biosphere have strong empirical support for the effects of CO₂ on plant growth (3) but limited empirical support for assumed effects on soil C accumulation.

Soil C stocks are determined by the balance between plant growth and the subsequent input of plant detritus to soil and C losses through microbial decomposition. Earth system models typically treat decomposition as a function of temperature and water content (1) and do not include direct effects of increased CO₂, which are generally assumed to be small (6). These models commonly describe soil C dynamics with first-order kinetics (7), such that C loss from soil increases proportionally with the size of the soil C pool (8) (Eq. 1). Thus, if increased CO₂ enhances C input to soil, these models project that decomposition will also increase. Consistent with these expectations, CO₂ enrichment is frequently found to increase CO₂ efflux from soil (9). However, in some cases this increase was greater than the first-order expectation (10, 11), suggesting that increased CO₂ may stimulate decomposition beyond what is expected from increased substrate availability alone. It is not known whether this disproportionate response to increased atmospheric CO₂ is general across ecosystems.

Disentangling C input to and loss from soil is possible in short-term laboratory experiments using isotopic tracers (10) and can be inferred from mass balance calculations (12). However, synthetic approaches disentangling C inputs and losses across broad data sets from the field have not been developed. Here, we use an approach that combines measurements of plant production, microbial respiration, and soil C stocks to estimate CO₂ effects on soil C turnover, thereby providing insight into the mechanisms determining long-term soil C storage (13). We derived effect sizes from a one-pool biogeochemical model of soil C cycling (8), the same form used by several global C cycling models (1, 2)

$$C_t = C_0[\exp(-kt)] + I/k[1 - \exp(-kt)] \quad (1)$$

where C_t is the soil C content (g C m⁻²) at time t (year); C_0 is the soil C content at the start of a CO₂ enrichment experiment (g C m⁻²); k is the decomposition rate constant (year⁻¹, the rate at which C leaves the soil system); and I is the annual input of C to soil (g C m⁻² year⁻¹). The model was constrained by multiple independent data streams from 53 CO₂ enrichment experiments (see table S1 and databases S1 to S4). We then used meta-analysis to synthesize the results (13).

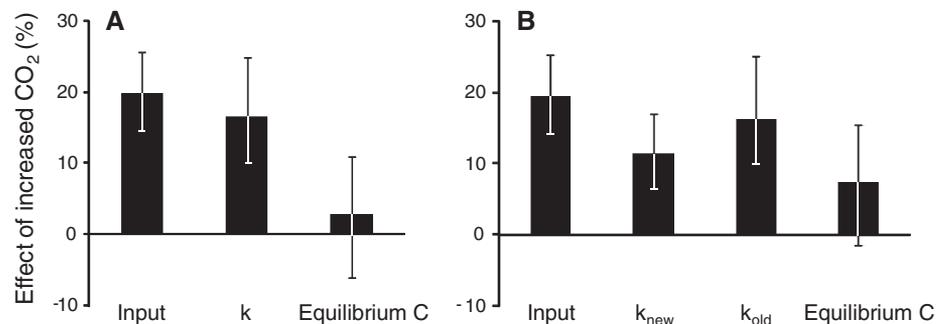
Using this approach, the data indicate that CO₂ enrichment increased soil C input (I in Eq. 1) by 19.8% (Fig. 1A), with lower responses in cropland (10.9%) compared with grassland (20.1%) and forest (23.3%) (table S2). These estimates are consistent with past syntheses of the responses of plant production to experimental CO₂ enrichment (3).

If the decomposition rate constant k of soil C in Eq. 1 were unaffected by CO₂ enrichment, then microbial decomposition would change proportionally with the size of the soil C pool. That is, C loss from decomposition would increase with CO₂ enrichment because the pool of soil C increased, not because the rate of decomposition (i.e., k) increased. However, our meta-analysis showed that CO₂ enrichment significantly increased k by 16.5% (Fig. 1A). The average effect

¹Center for Ecosystem Science and Society, Northern Arizona University, Flagstaff, AZ 86011, USA. ²Department of Botany, School of Natural Sciences, Trinity College Dublin, Dublin 2, Ireland. ³Department of Microbiology and Plant Biology, University of Oklahoma, Norman, OK 73019-0245, USA. ⁴Department of Biology, University of Florida, Gainesville, FL 32611-8525, USA. ⁵Center for Earth System Science, Tsinghua University, Beijing 100084, China. ⁶Department of Biological Sciences, Northern Arizona University, Flagstaff, AZ 86011, USA.

*Corresponding author. E-mail: cjvangroenigen@nau.edu

Fig. 1. Results of a meta-analysis on the response of soil C dynamics to increased levels of atmospheric CO₂. (A) The effect of increased CO₂ on soil C input (input), soil C turnover (k), and projected equilibrium soil C, based on a one-pool soil C model (Eq. 1). (B) The effect of increased CO₂ on soil C input, new soil C turnover (k_{new}), old soil C turnover (k_{old}), and projected equilibrium soil C, based on a two-pool soil C model (Eq. 2). Results are based on 53 experimental comparisons. Effect sizes were weighted by replication, adjusted by the number of comparisons per experimental site. All error bars represent 95% confidence intervals.



of increased CO₂ was similar in grasslands, forests, and croplands, although the effect in the latter category was not significant (see table S2). Correspondingly, increased atmospheric CO₂ significantly reduced soil residence time ($1/k$, the average time that a C atom spends in the soil) by 2.4 years (95% confidence interval, 1.5 to 3.8 years) (see table S3).

Why does increased atmospheric CO₂ stimulate the turnover of soil C? Larger soil C input rates under increased CO₂ cause an enhanced supply of easily metabolized substrates (I), which can stimulate the decomposition of native soil organic matter, mobilizing C reserves assumed to be protected from microbial attack (I). This process, commonly called “priming” (I), has previously been described as short-term and idiosyncratic (I). However, recent studies indicate that it is a widespread and persistent phenomenon (I , I). Our results now suggest that it is also a general and prolonged response to increased atmospheric CO₂.

Furthermore, atmospheric CO₂ enrichment can increase soil water content (I), due to improved efficiency of water use by plants, which reduces soil water loss through transpiration (I). Because low water availability limits the physiological performance of soil microbes and their access to substrate, a CO₂-induced increase in soil water content may increase decomposition rates in arid and semiarid ecosystems (I).

Our results are based on a one-pool soil C model, but two- or three-pool soil C models that distinguish labile and recalcitrant C pools may describe changes in soil C stocks over time more accurately (I); however, parameters in a conventional multipool soil C model could not be estimated with the available data (I). Thus, we took a slightly different approach and applied a model that distinguished the k of soil C initially present at each experimental site (k_{old}) from the k of soil C added after CO₂ enrichment started (k_{new}).

$$C_t = C_0[\exp(-k_{old} \times t)] + I/k_{new}[1 - \exp(-k_{new} \times t)] \quad (2)$$

Based on this approach, we found that CO₂ enrichment increased both k_{old} and k_{new} (Fig. 1B and table S4) to similar degrees. These findings corroborate recent studies suggesting that increased CO₂ stimulates the decomposition of old soil C (I , I) as well as new soil C pools (I).

Given enough time, soil C stocks will eventually reach equilibrium of I/k (Eq. 1) or I/k_{new} (Eq. 2), where soil C input through plant growth is balanced by soil C loss through decomposition. If increased CO₂ alters only I and not k , an assumption made in early Earth system models (I), predicted equilibrium soil C contents for the experiments in our database would increase to the same degree as soil C input

rates, that is, by 19.8%. However, when we take into account the effect of CO₂ on k , increased CO₂ does not significantly affect the predicted equilibrium soil C contents (Fig. 1, A and B). In other words, the observed increase in k largely negates the predicted soil C accumulation due to increased atmospheric CO₂. Although increased CO₂ stimulates plant growth and C input to soil, the microbial response counteracts this response, thereby reducing the net effect of rising CO₂ levels on soil C stocks.

The effect of increased CO₂ on I/k and I/k_{new} (which represent projected long-term responses) are quantitatively similar to the effect of increased CO₂ on soil C contents found in several recent meta-analyses focusing on short-term responses (I , I). This suggests that studies that are focused on processes (and thus allow estimation of I and k) and those focused on soil C storage provide complementary answers. Importantly, our approach allows the separation of inputs and losses and thus explains the apparent mismatch between the stimulation of plant growth and the limited response of soil C (I , I).

Current Earth system models include only some of the possible processes underlying the response of k to increased atmospheric CO₂. In most models, atmospheric CO₂ enrichment could influence decomposition by altering soil moisture (I). Furthermore, some Earth system models partition plant litter into multiple pools with different decomposition rates. Increased CO₂ could drive changes in k due to shifts in vegetation composition, because the ratio of these pools differs between vegetation types (I). However, because of their relative short duration, experiments in our data set did not allow for the large shifts in vegetation composition that affect soil C dynamics in these models (I). Therefore, this mechanism does not explain our findings. Alternatively, Earth system models that include a nitrogen (N) cycle can simulate changes in C to N ratio and decomposability of litter (I). However, because atmospheric CO₂ enrichment generally has little effect on litter quality (I), this mechanism probably plays a limited role in explaining the response of k . The priming effect itself is not explicitly represented in any Earth system model. Including priming in ecosystem models captures increased decomposition rates with increased atmospheric CO₂ (I), a result consistent with our findings. Future Earth system models might improve by incorporating the priming effect in their land surface mode as well.

References and Notes

1. P. Friedlingstein *et al.*, *J. Clim.* **19**, 3337–3353 (2006).
2. P. E. Thornton, J.-F. Lamarque, N. A. Rosenbloom, N. M. Mahowald, *Global Biogeochem. Cycles* **21**, GB4018 (2007).
3. E. A. Ainsworth, S. P. Long, *New Phytol.* **165**, 351–372 (2005).

4. B. A. Hungate, R. B. Jackson, C. B. Field, F. S. Chapin III, *Plant Soil* **187**, 135–145 (1995).
5. P. Smith, *Soil Use Manage.* **20**, 264–270 (2004).
6. R. J. Norby, M. F. Cotrufo, P. Ineson, E. G. O'Neill, J. G. Canadell, *Oecologia* **127**, 153–165 (2001).
7. W. J. Parton, D. S. Schimel, C. V. Cole, D. S. Ojima, *Soil Sci. Soc. Am. J.* **51**, 1173–1179 (1987).
8. J. S. Olson, *Ecology* **44**, 322–331 (1963).
9. D. R. Zak, K. S. Pregitzer, J. S. King, W. E. Holmes, *New Phytol.* **147**, 201–222 (2000).
10. E. Paterson *et al.*, *Soil Biol. Biochem.* **40**, 2434–2440 (2008).
11. K. M. Carney, B. A. Hungate, B. G. Drake, J. P. Megonigal, *Proc. Natl. Acad. Sci. U.S.A.* **104**, 4990–4995 (2007).
12. J. E. Drake *et al.*, *Ecol. Lett.* **14**, 349–357 (2011).
13. Materials and methods are available as supplementary materials on Science Online.
14. S. Fontaine *et al.*, *Nature* **450**, 277–280 (2007).
15. E. Blagodatskaya, Y. Kuzyakov, *Biol. Fertil. Soils* **45**, 115–131 (2008).
16. J. W. Dalenberg, G. Jager, *Soil Biol. Biochem.* **21**, 443–448 (1989).
17. F. A. Dijkstra, W. Cheng, *Ecol. Lett.* **10**, 1046–1053 (2007).
18. E. J. Sayer, M. S. Heard, H. K. Grant, T. R. Marthews, E. V. J. Tanner, *Nature Clim. Change* **1**, 304–307 (2011).
19. K. J. van Groenigen, C. W. Osenberg, B. A. Hungate, *Nature* **475**, 214–216 (2011).
20. S. D. Wullschlegel, T. J. Tschaplinski, R. J. Norby, *Plant Cell Environ.* **25**, 319–331 (2002).
21. E. Pendall *et al.*, *Global Biogeochem. Cycles* **17**, GB1046 (2003).
22. E. A. Davidson, S. E. Trumbore, R. Amundson, *Nature* **408**, 789–790 (2000).
23. R. P. Phillips *et al.*, *Ecol. Lett.* **15**, 1042–1049 (2012).
24. B. A. Hungate *et al.*, *Glob. Change Biol.* **15**, 2020–2034 (2009).
25. W. M. A. Sillen, W. I. J. Dieleman, *Biogeosciences* **9**, 2247–2258 (2012).
26. K. J. van Groenigen *et al.*, *Proc. Natl. Acad. Sci. U.S.A.* **103**, 6571–6574 (2006).
27. Y. Luo *et al.*, *Glob. Change Biol.* **17**, 843–854 (2011).
28. V. K. Arora *et al.*, *J. Clim.* **26**, 5289–5314 (2013).
29. N. Perveen *et al.*, *Glob. Change Biol.* **20**, 1174–1190 (2014).

Acknowledgments: This research was supported by the Biological and Environmental Research program, Office of Science, U.S. Department of Energy; the Western Regional Center of the National Institute for Climatic Change Research; and the Irish Research Council, cofunded by Marie Curie Actions under the Seventh Framework Program (FP7). X.Q. and Y.L. received financial support from the U.S. Department of Energy, Terrestrial Ecosystem Sciences grant DE SC0008270. We thank M. Hoosbeek, K. Stürner, A. Talhelm, and D. Zak for sharing unpublished data with us. Many thanks go to T. Xu for assisting us with writing the Matlab code. The data reported in this paper are available online as supplementary materials.

Supplementary Materials

www.sciencemag.org/content/344/6183/508/suppl/DC1
Materials and Methods
Figs. S1 to S3
Tables S1 to S4
Databases S1 to S6
References (30–106)

10 December 2013; accepted 4 April 2014
Published online 24 April 2014;
10.1126/science.1249534



Faster Decomposition Under Increased Atmospheric CO₂ Limits Soil Carbon Storage

Kees Jan van Groenigen *et al.*
Science **344**, 508 (2014);
DOI: 10.1126/science.1249534

This copy is for your personal, non-commercial use only.

If you wish to distribute this article to others, you can order high-quality copies for your colleagues, clients, or customers by [clicking here](#).

Permission to republish or repurpose articles or portions of articles can be obtained by following the guidelines [here](#).

The following resources related to this article are available online at www.sciencemag.org (this information is current as of March 31, 2016):

Updated information and services, including high-resolution figures, can be found in the online version of this article at:

</content/344/6183/508.full.html>

Supporting Online Material can be found at:

</content/suppl/2014/04/23/science.1249534.DC1.html>

This article **cites 101 articles**, 7 of which can be accessed free:

</content/344/6183/508.full.html#ref-list-1>

This article has been **cited by 1** articles hosted by HighWire Press; see:

</content/344/6183/508.full.html#related-urls>

This article appears in the following **subject collections**:

Atmospheric Science

</cgi/collection/atmos>

Geochemistry, Geophysics

/cgi/collection/geochem_phys